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Student Exploration- Calorimetry Lab (ANSWER KEY) Gizmo Warm-up A calorimeter is an insulated container filled with a liquid, usually water When a hot object is placed in the calorimeter, heat energy is transferred from the object to the water and the water

### **[Book] Answer Key For Calorimetry Lab**

the initial temperature of 20 °C to result in a final system temperature of 20.45 °C. (Hint: Start by calculating how much heat energy is needed to change the water's temperature by 0.45 °C). Show your work. Use the Gizmo to check your answer.  $3000\text{g} \times 4.184 \text{ j/g} \times 0.45 \text{ c} = 5,648\text{j}$   $\times .795 \text{ j/g} \times -59.55\text{C} = - 5 ,648$  Mass of granite =  119.31g 5. Extend your thinking: In addition to calculating ...

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