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Chapter 4 Atoms Practice Problems 1) The primary substances of which all other things are composed are A) molecules. B) compounds. C) elements. D) electrons. E) protons. 2) Which of the following is a characteristic of the modern periodic table? A) A group is a horizontal row on the periodic table. B) A period is a column on the periodic table.

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Chapter 4 Atomic Structure 33 SECTION 4.1 DEFINING THE ATOM (pages 101–103) This section describes early atomic theories of matter and provides ways to understand the tiny size of individual atoms. Early Models of the Atom (pages 101–102) 1. Democritus, who lived in Greece during the fourth century B.C.,

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26. The atomic number of carbon is 6. The atomic number of nitrogen is 7. The atomic number of oxygen is 8. Name the isotope represented by the drawing. 27. Why were the proton and electron discovered before the neutron? 28. Explain why a neutral atom cannot have one proton, one neutron, and two electrons. 29.

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## **Chapter 4 - Atomic Structure - Mr. Pieratt's Chemistry Lab**

104 Chapter 4 • The Structure of the Atom ■ Figure 4.2 In his book A New System of Chemical Philosophy, John Dalton presented his symbols for the elements known at that time and their possible combinations.

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