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Objective: Describe the
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Vocabulary:

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Chemistry of Life |

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Biology I

Biology Chemistry Key

Study Guide. STUDY.
PLAY. Atom. building
block of matter.

Element. pure
substance; cannot be
broken into other
substances by physical
or chemical means.

Ion. atoms having lost
or gained electrons.

Now have a charge.

Isotope. atoms of the
same element having
different number of
neutrons. Different

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of matter. 6.1 Atoms,
Elements, and
Compounds Atoms are
the building blocks of
matter. Chapter 6
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Neutrons and protons
are located at the

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center of the atom.

Protons are positively charged particles.

Neutrons are particles that have no charge.

Chapter 6 Chemistry In Biology Study Guide Answers

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good understanding of algebra is helpful. Did you know that everything is made out of chemicals?

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Chapter 2 The
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Guide Figure 2.1 Foods
such as bread, fruit,
and cheese are rich
sources of biological
macromolecules. The
elements carbon,
hydrogen, nitrogen,
oxygen, sulfur, and
phosphorus are the key
building blocks of the
chemicals found in
living things.

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Covalent Bond. electrons are shared between atoms. Ionic Bond. electrons are transferred from one atom to another and the two atoms are held together by the attraction of opposite charges. Hydrogen Bond. attraction between hydrogen atom on one water molecule and oxygen atom on another water molecule - not as strong as covalent or

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