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13.78 kPa 6. Find the partial pressure of carbon dioxide in a gas mixture with a total pressure of 30.4 kPa if the partial pressures of the other two gases in the mixture are 16.5 kPa and 3.7 kPa. 30.4

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States of Matter

Notice in particular the curves for NH_4NO_3 and CaCl_2 . The dissolution of ammonium nitrate in water is endothermic ($\Delta H_{\text{soln}} = +25.7 \text{ kJ/mol}$), whereas the dissolution of calcium chloride is exothermic ($\Delta H_{\text{soln}} = -68.2 \text{ kJ/mol}$), yet Figure 13.4.1 shows that the solubility of both compounds increases sharply with increasing temperature. In fact, the magnitudes of the changes in both enthalpy ...

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is constant in the following problems 11

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A sample of air in a syringe exerts a pressure of 102 atm at 220°C The

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