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E–Z configuration, or the E–Z convention, is the IUPAC preferred method of describing the absolute stereochemistry of double bonds in organic chemistry. It is an extension of cis – trans isomer notation (which only describes relative stereochemistry) that can be used to describe double bonds having two, three or four substituents.

E-Z notation - Wikipedia

The E-Z system gets around this problem completely - but unfortunately makes things slightly more difficult for the simple examples you usually meet in introductory courses. How the E-Z system works We'll use the last two compounds as an example to explain how the system works.

E-Z notation for geometric isomerism - chemguide

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So the E-Z system is more inclusive than the cis-trans terminology, so E-Z is often a better way to come up with the configuration of a double bond. Let's assign a configuration to this double bond, and let's start with the carbon on the left side. We look at the atoms directly bonded to that carbon. There's a bromine, and there's a chlorine.

E-Z system (video) | Alkene nomenclature | Khan Academy

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