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the electrolytic cell,
flow of electrons is
from. A. Cathode to
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20 Questions MCQ Test

AP Chemistry:
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Answers 14. Questions
14-17 The spontaneous
reaction that occurs
when the cell in the
picture operates is as
follows: $2\text{Ag}^+ + \text{Cd (s)}$
 $\rightarrow 2 \text{Ag (s)} + \text{Cd}^{2+}$ (A)
Voltage increases. (B)
Voltage decreases but
remains $> \text{zero}$.

AP Chemistry:
Page 8/28

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NEET Chemistry :

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Questions. 31. In acidic

medium MnO_4^{2-} is

converted to Mn^{2+} .

The quantity of

electricity in faraday

required to reduce 0.5

mole of MnO_4^{2-} to Mn^{2+}

would be : 2.5. 1.

5. 0.5. Answer. 32.

When Cu reacts with

AgNO_3 solution, the

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reaction takes place is:

Questions

Important Questions of Electrochemistry for NEET Chemistry

...

APEF -

Electrochemistry -
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Questions A 1. Which
of these chemical
reactions is an

oxidation-reduction

reaction? A. $\text{Fe} + \text{S} \hat{=}$

FeS C. $\text{AgNO}_3 + \text{NaCl} \hat{=}$

$\text{AgCl} + \text{NaNO}_3$ B. CO_2

$+ \text{H}_2\text{O} \hat{=}$ H_2CO_3 D.

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$\text{H}_2\text{SO}_4 + 2\text{NaOH} \rightarrow$

$\text{Na}_2\text{SO}_4 + 2\text{H}_2\text{O}$ C 2.

What happens to the oxidizing agent in an oxidation-reduction reaction? A. It is oxidized as it gains electrons. C.

APEF - Electrochemistry - Multiple Choice Questions ...

A current is passed through two voltameters connected in series. The first

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voltmeter connected in series. The first voltmeter contains XSO_4 (aq) while the second voltmeter contains Y_2SO_4 (aq). The relative atomic masses of X and Y are in the ratio of 2:1. The ratio of the mass of X liberated to the mass of Y liberated is:

NEET Chemistry Electrochemistry Questions Solved

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Questions 1. In an electrolytic cell the electrode at which the electrons enter the solution is called the _____ ; the chemical change that occurs at this electrode is called _____.

Sample Questions - Chapter 21

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SAHOTA 03

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- Page 1 of 22 1. DO
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in this booklet. These
are actual Provincial
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THE. URGE

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**THE "OFFICIAL"
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MULTIPLE CHOICE.

Choose the one alternative that best completes the statement or answers the question. 1) The gain of electrons by an element is called_____.

- A)oxidation
- B)sublimation
- C)reduction
- D)disproportionation
- E)fractionation 2) _____

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is reduced in the following reaction:
$$\text{Cr}_2\text{O}_7^{2-} + 6\text{S}_2\text{O}_3^{2-} + 14\text{H}^+ \rightarrow 2\text{Cr}^{3+} + 3\text{S}_4\text{O}_6^{2-} + 7\text{H}_2\text{O}$$

**A.P. Chemistry
Practice Test - Ch.
17: Electrochemistry
A ...**

Multiple choice questions. Try the following multiple choice questions to test your knowledge of this chapter. For each question there is one

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Multiple Choice
Questions
Answers

correct answer. The periodic table, physical constants and relative atomic masses needed for these problems are given on the inside covers of Chemistry, fourth edition by C.E. Housecroft and E.C. Constable. Once you have answered the questions, click on ...

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Multiple Choice.
Identify the choice that
best completes the
statement or answers

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the question. 1.
The half-reaction that occurs at the cathode during the electrolysis of molten sodium bromide is .

AP Chemistry- Electrochemistry

Examples of Multiple Choice Questions from GENERAL CHEMISTRY. Choose your chapter:
Fundamentals of Chemistry | Chemical Formulas & Composition

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Reactions | | Atomic
Structure | Chemical
Periodicity | Chemical
Bonding | Molecular
Structure/Covalent
Bonding Theories |
Molecular Orbital
Theory |

**Multiple Choice
Questions - Texas
A&M University**
Electrochemistry Unit

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Review Area 1 -

Multiple Choice Use the following information to answer the next question Over 100 years ago, Gustave Eiffel designed the support structure for the Statue of Liberty. An iron framework was constructed and overlaid with copper sheets in such a way that the copper did not come into direct contact with the iron.

1. ... *Page 22/28*

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