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IB Chemistry Exam Secrets Topic 7: Equilibrium. This unit has the smallest allocation of marks on both paper 1 and paper 2. However, it is easily integrated into any stoichiometry question. Since the topic is so small it is very easy to focus on the few required understandings and score well.

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Topic 7: Equilibrium - IB Exam Preparation - Studynova

IB Chemistry (Ellesmere College) 7.2
Past Paper Questions. 1. $I_2(g) + 3Cl_2(g) \rightleftharpoons 2ICl_3(g)$... State and explain the effect on the value of the equilibrium constant when the temperature of the reaction is increased. (2) (c) State and explain the colour change when hydrochloric acid is

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added to the reaction solution at equilibrium. ...

IB Chemistry (Ellesmere College) - 7.2 Past Paper Questions

IB chemistry notes on equilibrium (sl) ...
The following notes were written for the previous IB syllabus (2009). The new IB syllabus for first examinations 2016 can

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be accessed by clicking the link below.
IB syllabus for first examinations 2016.
Equilibrium (sl) 8.1 - Dynamic
equilibrium:

IB Chemistry notes: equilibrium (sl)
Homogenous equilibrium: reactants and
products are present in one phase.
(Common one is reactions in gaseous

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phase) Heterogeneous equilibrium:
reactants and products exist in more
than one phase; K_c of the reverse
reaction is the same as $1/K_c$ of the
forward reaction $K_c(\text{reverse}) = 1/K_c$
(forward) Effect of conditions on
equilibrium constant ...

7.1 Equilibrium - IB Alchemy

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Chem12 Equilibrium : Exam Questions
1-50 14) Gas X₂ reacts with gas Y₂ to
give gaseous XY according to the
equation $X_2 + Y_2 \rightleftharpoons 2XY$ $K_{eq} =$
 26×10^{-3} What is K_{eq} for the reverse
reaction? a) 68×10^{-6} b) 51×10^{-2} c)

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Organic Chemistry; Topic 11
Measurement and analysis; IB Chemistry
HL Notes.

IB Past Papers | IB Chemistry Tutors

IB Chemistry Review Detailed notes for
IB Chemistry course. Topic 7: Equilibrium
Use this outline in conjunction with the
IB syllabus. ... In the dynamic

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equilibrium, the concentration of the products and reactants remains constant although the reaction never goes to completion.

IB Chemistry Review: Topic 7: Equilibrium

Topic 7 Equilibrium ; Materials; Topic 7
SL Past Papers Questions & Answers ...

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Topic 8 HL Past Papers Questions &
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www.SmashingScience.org**

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IB Chemistry Answers was a wonderful learning tool for understanding Science. Jash Parekh As a Science Instructor when I started my carrier for the first time at ABWA School, I found this resource incredibly helpful for me and my students and I highly recommend it to anyone using IB Chemistry Answers website.

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IB Chemistry - IB dead

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Chemistry Standard level Paper 1 - IB Documents

Topic 9 Oxidation and reduction (IB Chemistry) The Reactivity Series of Metals, Corrosion and Redox Reaction Notes - The reactions of metals with

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oxygen, water and acids, displacement reactions, rusting of iron and preventing its corrosion, plus an introduction to oxidation-reduction (redox) reactions, simple cells etc. The notes on each metal are listed in reactivity order.

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IB Chemistry - Mr Weng

A short revision podcast on IB SL
Chemistry Topic 7, Equilibrium.

Answering some basic questions on the
topic. (Recorded with [http://screencast-o-
matic.com](http://screencast-o-matic.com))

Topic 7: Equilibrium Questions (IB SL

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Chemistry)

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What will the effect be on the position of

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equilibrium and the K_c value if the pressure is decreased on the reaction vessel? A. K_c decreases and equilibrium shifts to the left. B. K_c increases and equilibrium shifts to the left. C. K_c is unchanged but equilibrium shifts to the right. D. K_c is unchanged but equilibrium shifts to the left.

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Equilibrium Quiz - IB Exam Preparation - Studynova

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