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Chemistry IF8766 4 ©Instructional Fair, Inc ISOTOPES AND AVERAGE Name ATOMIC MASS Elements come in a variety of isotopes, meaning they are made up of atoms with the same atomic number but different atomic masses These atoms differ in the number

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Chemistry 301

enstructional Fair, Inc. 5. 35 g of $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$ Determine the number of grams in each of the quantities below. 1, 2.5 moles of NaCl 2. 0.50 moles of 3, 1.70 moles of KMnO_4 4, 0.25 moles of KCl e, ST 5. 3.2 moles of $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$ Chemistry IF8766

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Chemistry IF8766 82 ©Instructional Fair, Inc LE CHATELIER'S PRINCIPLE Name _ Le Chatelier's Principle states that when a system at equilibrium is subjected to a stress, the system will shift its equilibrium point in order to relieve the stress Stoichiometry Mass Problems Answers Chemistry If8766

[MOBI] Answers To Chemistry If8766

Chemistry IF8766 ©Instructional Fair, Inc. ATOMIC STRUCTURE AND THE PERIODIC TABLE CHAPTER 6 & 7 WORKSHEET Read the clue and give the element symbol that applies. CLUE 1. 2.

|He SAI

A chemical equation is the symbol in Chemistry that represents chemical reaction with the help of chemical formulas. It contains the chemical

substances that are involved in the reaction. It contains reactants and products. The reactants are the elements that react with one and another in a chemical reaction, while the products are the elements ...

49 Balancing Chemical Equations Worksheets [with Answers]

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Chemistry IF8766 Equilibrium Shift right + OH-(aq) Equilibrium Shift none Same 10,6 kcal Amount NaOH(s) Same Same 82 decreases same Same remains the increases same same same Same Sawte same same same (Remember that pure solids and liquids do not affect equilibrium values.) [Na+] same Sa [OH-] same Same same Sa me same same sanqe

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