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be explained by solving at the quantum level that same Kepler problem, solving which at the classical level allowed one to understand the solar system.

Introduction to Quantum Mechanics

CHAPTER 2. INTRODUCTION TO QUANTUM MECHANICS 24 An important example of operators on C^2 are the Pauli matrices, $\sigma_0 \equiv I \equiv \begin{pmatrix} 1 & 0 \\ 0 & 1 \end{pmatrix}$, $\sigma_1 \equiv X \equiv \begin{pmatrix} 0 & 1 \\ 1 & 0 \end{pmatrix}$, $\sigma_2 \equiv Y \equiv \begin{pmatrix} 0 & -i \\ i & 0 \end{pmatrix}$, $\sigma_3 \equiv Z \equiv \begin{pmatrix} 1 & 0 \\ 0 & -1 \end{pmatrix}$. (2.19) The Pauli matrices are related to each other through commutation rela-

Chapter 2 Introduction to Quantum Mechanics

quantum mechanics in a more systematic fashion in Chapter 4. Quantum mechanics is used to examine the motion of a single particle in one dimension, many particles in one

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description of the state of a quantum object (or system) is given mathematically by the state vector $|\psi\rangle$ (or wavefunction $\psi(\mathbf{r}, t)$). The situation is however different than in classical mechanics.

Lecture notes, Chapter 2.

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The s sub-level has one electron orbital, therefore, s can hold a maximum of two electrons. The p sublevel has three electron orbitals, each one can hold up to two, so we have six maximum in the p sublevel. Also remember that each one of these sub-levels has a shape distinct to them.

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