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2 Chemistry, First Canadian Edition 1.17 An organic compound consists of molecules made from carbon, hydrogen and oxygen atoms. There can be other atoms - for example, nitrogen, sulfur. 1.19 1. Acetylsalicylic acid, the common mild pain reliever, is a derivative of the natural product, salicin, obtained from willow bark. 2.

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EQUILIBRIUM (Nelson Chemistry 12, Section 7.1, Pg.424) □ Some reactions occur in only one direction □ Frying an egg □ $2 \text{Pb}(\text{NO}_3)_2 \rightarrow 2 \text{PbO} + 4 \text{NO}_2 + \text{O}_2$ □ Burning a piece of paper □ $\text{C}_6\text{H}_{12}\text{O}_6(\text{s}) + 6 \text{O}_2(\text{g}) \rightarrow 6 \text{CO}_2(\text{g}) + 6 \text{H}_2\text{O}(\text{l})$ □ Other reactions can occur in both directions (reversible) □ $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}(\text{s}) \leftrightarrow \text{CuSO}_4(\text{s}) + 5 \text{H}_2\text{O}(\text{l})$ □ $\text{H}_2(\text{g}) + \text{I}_2(\text{g}) \leftrightarrow 2 \text{HI}(\text{g})$ □ $\text{N}_2\text{O}_4(\text{g}) \leftrightarrow 2 \text{NO}_2(\text{g})$ EQUILIBRIUM: when the forward and reverse reactions are occurring at the same rate.

EQUILIBRIUM (Nelson Chemistry 12, Section 7.1 ...

$n = 12.5 \text{ mmol}$ Solution: $\text{H}^+(\text{aq})$ and $\text{NH}_3(\text{aq})$ react in a 1:1 ratio, so $\text{NH}_3 n = 12.5 \text{ mmol}$ $3 \text{ NH}(\text{aq})$ $3 \text{ NH}(\text{aq})$ $3 [\text{NH}(\text{aq})]$ 12.5 mmol 40.0 mL 0.313 mmol/mL $[\text{NH}(\text{aq})]$ $0.31 \text{ mol/L} = = = = n V(\text{b})$ Given: $[\text{NH}_4^+(\text{aq})] = 0.31 \text{ mol/L}$; $K_b = 1.8 \times 10^{-5}$ Required: pH Analysis: $\text{NH}_4^+(\text{aq}) \rightleftharpoons \text{NH}_3(\text{aq}) + \text{H}^+(\text{aq})$

Section 8.7: Acid-Base Titration Tutorial 1 Practice, page 547

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Here are the answers for all of the class handouts. I don't guarantee that they are all correct, so look at them critically. Organic Nomenclature. Quantum. Energy and Rates. Organic Reactions. Organic Assignment. Equilibrium. Ksp. Acids and Bases. Redox.

SCH4U Solutions - Science, Chemistry and Physics at FHCI

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