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A useful thing to note is that the sum of the mole fractions of all substances in a mixture equals 1. Thus the mole fraction of C 6 H 6 in Example  $\{\{1\}\}$  could be calculated by evaluating the definition of mole fraction a second time, or-because there are only two substances in this particular mixture-we can subtract the mole fraction of the C 10 H 8 from 1 to get the mole fraction ...

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